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Brønsted-Lowry Acids & Bases
Identify each species in the

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following equation as either the Brønsted-Lowry acid, the Brønsted-Lowry base, the conjugate acid, or the conjugate base. Identify the conjugate acid-base pairs in the reaction.

$$\text{H}_2\text{SO}_4(\text{aq}) + \text{HPO}_4^{2-}(\text{aq}) \rightarrow \text{HSO}_4^{-}(\text{aq}) + \text{H}_2\text{PO}_4^{-}(\text{aq})$$

Strong vs.

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Weak Acids and Bases Strong
Acids:

Chapter 15: Acids and Bases Acids
and Bases

A strong acid is an acid which will
dissociate completely to form an
anion and a hydrogen ion. In

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contrast, a base is any type of molecule which can accept hydrogen ions from the solution.

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The Nature of Acid and Bases.
Arrhenius definition says the an
acid supplies _____ to an aqueous
solution. An Arrhenius bases
supplies _____ to an aqueous
solution. Bronsted- Lowry concept

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Workbook
says that an acid is a _____

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a base is a _____

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Identify the acid and base and the
conjugate pairs in the previous
equations. Conjugate Pairs.

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Conjugate pairs differ by one H^+ in their formulas . Write dissociation reactions for each of the following acids in water and ID conjugate acid – base pairs. formic acid ($HCOOH$) perchloric acid.

Equilibrium Expressions for Acid Dissociations. Write an equilibrium

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Workbook
expression for each of the
equations above. a) b)

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“ ACIDS AND BASES ” NAME:
_____ Šimon Mráz Study Goals •

Describe the characteristics of

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Arrhenius, Brønsted-Lowry, and organic acids and bases. • Identify conjugate acid-base pairs in Brønsted-Lowry acids and bases. • Write equations for the dissociation of strong and weak acids and identify the direction of the reaction. • Write the

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expression for the dissociation constants of a weak acid or weak base.

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SAHOTA 03 Acid Base Study
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THE “OFFICIAL ” CHEMISTRY 12 ACID BASE STUDY GUIDE

Acids and Bases An object ' s pH level can be tested using indicators. Objects with a low pH are acids, and those with a high pH are bases. Acids and bases react together to form water and salt.

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Workbook
Brønsted-Lowry definition of acids and bases because acids are chemicals that donate protons and bases are ones that accept protons. I then pass out the Acids and Bases Reading Questions which guide students in finding the background knowledge I want

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Workbooks for students to learn for this unit from their text. I note that they should work on questions 1-19 and also work with a partner to study their acid and base names.

Eleventh grade Lesson
Introduction to Acid Base Theory

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Look at Figure 12 – 10 on page 297.

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List in order, beginning with DNA, the levels of organization of eukaryotic DNA to form chromosomes.

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